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Original Paper

Two-step hydrolysis of Japanese cedar as treated by semi-flow hot-compressed water

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1 **Abstract**

2 Two-step hydrolysis of Japanese cedar (*Cryptomeria japonica*) was studied as treated
3 by semi-flow hot-compressed water at 230°C/10MPa for 15 min and 280°C/10MPa for 30
4 min as the first and second stages, respectively. As a result, at the first stage, hemicelluloses
5 and para-crystalline cellulose whose crystalline structure is somewhat disordered were found
6 to be selectively hydrolyzed as well as lignin decomposition, while crystalline cellulose at the
7 second stage. Totally 87.76% of Japanese cedar could be liquefied by the hot-compressed
8 water and were primarily recovered as various hydrolyzed products, dehydrated, fragmented
9 and isomerized compounds, as well as, organic acids in the water-soluble portion. The rest of
10 12.24% could not be hydrolyzed and remained as the water-insoluble residue composed
11 entirely of lignin. Based on the distributions of various products from hemicelluloses in
12 Japanese cedar, their decomposition pathways were independently proposed.

13

1 Introduction

2 In an attempt to alleviate both of the environmental and energy security problems,
3 ethanol produced from various biomass has long been proposed as a promising biofuel for
4 several decades¹. Lignocellulosics can be utilized to obtain fermentable saccharides for the
5 bioethanol production. However, unlike sugar and starch, the development of bioethanol
6 production from them has been impeded by their great resistance to hydrolysis due to the
7 cellulose crystallinity and presence of lignin. Therefore, various biological, chemical and/or
8 physical treatments such as enzymatic saccharification, acid hydrolysis, alkali process, steam
9 explosion, and supercritical and subcritical treatments have been developed²⁻⁷.

10 However, one drawback of using acid or alkali is the necessity for neutralization and
11 separation². In addition, Aida et al.⁸ elucidated that an increase in water temperature
12 (350-400°C) and pressure (40-70MPa) did not only heighten the dehydration of glucose to
13 5-hydroxymethylfurfural (5-HMF), but also enhanced the formation of 1,2,4-benzenetriol
14 (BTO), resulting in a greater loss of saccharides and inhibition effect in the subsequent
15 fermentability. Although the liquefaction rate became extremely high as treated cellulose by
16 subcritical water, it was found that the hydrolysis rate of cellulose was much lower than the
17 decomposition rate of glucose and its oligomers⁹. The economic and environmental
18 constraints, moreover, limit the applicability of these known methods. Thus, hot-compressed
19 water, milder conditions with high ionic product, seems to be a promising alternative as both

1 environmentally friendly solvent and attractive reaction media for a variety of applications¹⁰⁻¹¹.
2 Hot-compressed water refers to the water at supercritical or subcritical state, or at sufficiently
3 high pressure and temperature¹².

4 Besides, it has long been realized that the hydrolysis of cellulose and hemicelluloses
5 could not be optimized at the same treatment severity^{2,6,13-14}. Therefore, two-step treatment,
6 with the first stage performed at low severity to hydrolyze the hemicelluloses and the second
7 stage in which the solid residue from the first stage is consecutively treated, but at a higher
8 degree of severity, has gained a great amount of attention as an alternative to increase
9 saccharide yields^{6,13-14}.

10 Ando et al.¹⁰ investigated the decomposition behavior of bamboo, chinquapin and
11 Japanese cedar as treated by two-step hot-compressed water at 180°C for 20 min and 285°C
12 for 7 min using the semi-flow-type reactor with a flow-rate of 10 mL/min under 9.8MPa. In
13 this report, however, quantitative assessments of hydrolyzed saccharides and various products
14 derived from hemicelluloses and cellulose, together with a discussion of their decomposition
15 pathways, have not been sufficiently clarified.

16 In our previous work¹⁵, two-step hydrolysis of Japanese beech (*Fagus crenata*) as one
17 of the hardwoods treated by semi-flow hot-compressed water has been conducted. As a result,
18 hemicelluloses and cellulose were found to be independently hydrolyzed in the first and
19 second stages of the treatment, respectively. In order to gain better insights into the two-step

hot-compressed water treatment of various lignocellulosics, in this work, therefore, two-step hydrolysis of Japanese cedar (*Cryptomeria japonica*) as one of the softwoods treated by semi-flow hot-compressed water was investigated. Not only qualitative assessments of various hydrolyzed products, dehydrated, fragmented and isomerized compounds as well as organic acids recovered in the water-soluble portion, but also the quantification of these products were reported. Based on this line of informative evaluations, the independent decomposition pathways of hemicelluloses and cellulose were eventually proposed.

Materials and methods

Material and chemicals

Extractive-free wood flour of Japanese cedar (*Cryptomeria japonica*) was prepared^{15,16}. The weight percentages of products were calculated on the oven-dried extractive-free wood flour basis. Table 1 shows the chemical compositions of the extractive-free Japanese cedar, determined by Klason lignin determination method¹⁷ for lignin and the obtained acid hydrolysate for hemicelluloses and cellulose. All chemicals used in this study were of reagent grade without purification.

Hot-compressed water treatment and fractionation of the treated samples

The semi-flow hot-compressed water biomass conversion system and operational

procedures as described previously¹⁵ were used in this study. Based on our preliminary experiments¹⁶ to explore the relationship between treatment temperature under pressurized conditions (10MPa) and the degree of hydrolysis of hemicelluloses and cellulose, the treatment temperatures applied for the two-step treatment were set at 230 and 280°C, 10 MPa at the flow-rate of 10.0 mL/min. The hot-compressed water-soluble portion was collected every 1 min by the fraction collector. After settling in the ambient temperature and pressure for 12 hours, water-soluble portion was obtained by filtration with the 0.45-μm membrane prior to subsequent analyses. On the other hand, water-insoluble residue left over the reaction cell after the treatment was collected, oven-dried and studied on its chemical compositions.

Analytical methods

The analyses of the water-soluble portion were made by the high-performance anion-exchange chromatography (HPAEC), high-performance liquid chromatography (HPLC), gas chromatography-mass spectrometry (GC-MS) and capillary electrophoresis (CE). The same analysis systems and operating conditions were applied in accordance with our previous paper¹⁵.

Post-hydrolysis by dilute sulfuric acid was performed to estimate all recovered oligosaccharides in the water-soluble portion in terms of acid-hydrolyzed monosaccharides¹⁸. The obtained acid hydrolysate was neutralized with barium hydroxide, Ba(OH)₂, and then

1 filtrated by 0.45-μm membrane filter before HPLC analysis using the HPX-87P column.
2 Distilled water was used as an eluent, and the system operated at the flow-rate and
3 temperature of 0.6 mL/min and 85°C, respectively. The amount of the oligosaccharides was
4 then calculated based on Eq. 1 as follows:

$$\begin{aligned} \text{Oligosaccharides (wt\%)} &= \text{Total monosaccharide in acid hydrolysate (wt\%)} - \\ &\quad \text{Monosaccharide in water-soluble portion (wt\%)} \quad (\text{Eq. 1}) \end{aligned}$$

9 Quantification of products in water-soluble portion

10 All the product percentages on the oven-dried weight of original wood flour were
11 calculated by using the peak area¹⁵ of chromatograms obtained from the HPAEC, HPLC, CE
12 and GC-MS, as shown in Figs. 1-6. All the plots of product percentages shown in the figures
13 were not exactly based on the real scale. Some of them were shifted by additional surpluses to
14 make a clear perception of each production trend.

16 Results and discussion

17 Hydrolysis of major cell wall components by hot-compressed water

18 As treated by the two-step hot-compressed water (230°C/10MPa/15min and
19 280°C/10MPa/30min), 87.76% of extractive-free Japanese cedar wood flour was liquefied

and primarily recovered as various compounds in the water-soluble portion, while the rest of 12.24% was not hydrolyzed and remained as the water-insoluble residue composed entirely of lignin, which was confirmed by Klason lignin determination. For the obtained water-soluble portion, the produced compounds were studied.

As seen in the temperature profile in Fig. 1, two-step treatment was performed over the wood flour of Japanese cedar. The dash lines indicate the transient time during an increase in temperature, while the solid lines depict the first and second stages of treatment at 230°C for 15 min and 280°C for 30 min, respectively, as shown in the real temperature profile. The hydrolyzed products obtained from hemicelluloses and cellulose are shown.

In the first stage (230°C/10MPa/15min), glucomanno-saccharides which include mannose, glucose and oligomeric glucomannan, galactose and acetic acid were produced. Besides, xylo-saccharides which include xylose and xylo-oligosaccharides, such as xylobiose, xylotriose, xylotetraose, xylopentaose, xylohexaose and the molecules with higher degree of polymerization (DP), arabinose and glucuronic acid were also recovered at this stage. In case of the production of acetic acid and glucuronic acid, since the amounts are relatively small compared to the others in Fig. 1, more apparent production trends of them can be seen later in Fig. 8. Therefore, these hydrolyzed products must originate from the major and minor hemicelluloses of the softwood, *O*-acetyl-galactoglucomannan and arabino-4-*O*-methylglucuronoxylan^{14,19-20}, respectively.

Based on the evidence achieved above, *O*-acetyl-galactoglucomannan can be 0.3:1:1.7 in the molar ratio of galactose:glucose:mannose with an acetyl residue per 48 monosaccharide units of the glucomannan backbone, while arabino-4-*O*-methyl glucuronoxylan can be 1:6.4 in the molar ratio of arabinose:xylose with a glucuronic acid residue per 118 monosaccharide units of the xylan backbone. These results are in agreement with those in the literature to some extent, however, the numbers of acetyl and acid residues are relatively smaller than the ones reported previously¹⁹. This might be due perhaps to the fact that they were not fully recovered in those forms.

Fig. 2 depicts various monosaccharides produced from Japanese cedar. In the first stage, mannose, galactose, xylose and arabinose were found. However, it is obvious that the amounts of mannose and xylose are much lesser than those of glucomanno-saccharides and xylo-saccharides (Fig. 1), indicating that most of those hydrolyzed saccharides were recovered as in the oligomeric forms.

Moreover in Fig. 2, a small amount of mannose was observed not only at the first stage shown by an arrow but also at the second stage shown by the inserted figure. Since hemicelluloses are supposed to be hydrolyzed in the first stage, the observed small peak of mannose and fructose at the second stage must be from cellulose. In addition, mannose and fructose are known as isomerized products of glucose in subcritical and supercritical water treatments²¹⁻²². This isomerization to mannose and fructose was ensured by the treatment of

1 D-glucose under the same hot-compressed water conditions. Such an isomerization of glucose
2 was also observed in our previous work on Japanese beech¹⁵.

3 On cello-saccharides which include glucose and cello-oligosaccharides, such as
4 cellobiose, cellotriose, cellotetraose, cellopentaose, cellohexaose and higher DP molecules, as
5 shown in Fig. 1, they were produced from 10 to 55 min in retention time. However, the one in
6 the first stage (10-25 min) would not be from crystalline cellulose because in the preliminary
7 experiment the crystalline structure of cellulose remained unchanged in such a temperature
8 around 230°C¹⁶. Therefore, the cello-saccharides just between the first and second stages
9 (10-25min) could be from para-crystalline cellulose, whose crystalline structure is somewhat
10 disordered²³.

11 In contrast to hardwoods, the dominating hemicellulose in softwoods is glucomannan
12 which can produce a significant amount of glucose as the hydrolyzed product. However, there
13 was a production of glucose and its oligomers also from the hydrolysis of para-crystalline
14 cellulose, leading to indistinguishable sources of them in the first stage. Since most of the
15 hydrolyzed saccharides from glucomannan were recovered in the form of oligomeric
16 glucomannan as described before, the glucose from glucomannan is assumed to be negligible.
17 Moreover, upon the assumption that a single glucose residue links to the mannose residue
18 evenly in the native glucomannan backbone, therefore, the oligomeric glucomannan can be
19 differentiated from the cello-oligosaccharides such as cellobiose, cellotriose, cellotetraose,

1 cellopentaose and cellohexaose. Thus, those cello-oligosaccharides (DP1~5) detected by the
2 HPAEC analysis, can be assumed to be exclusively from para-crystalline cellulose.
3 Consequently, the amount of glucose derived from oligomeric glucomannan could then be
4 estimated by the following equation.

$$\begin{aligned} & \text{Glucose from oligomeric glucomannan (wt\%)} = \\ & \quad \frac{\text{Total glucose in acid hydrolysate (wt\%)} - \text{Cello-oligosaccharides}}{\text{(DP1~5) in water-soluble portion (wt\%)}} \quad (\text{Eq. 2}) \end{aligned}$$

5
6
7
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10 The hemicelluloses were assumed to be completely hydrolyzed in the first stage. On the
11 other hand, cello-saccharides produced in the second stage were only from cellulose and thus
12 determined ordinarily. By this line of calculations, the glucomanno-saccharides and
13 cello-saccharides could be computed and plotted as shown in Fig. 1.

14 From the results as in Figs. 1 and 2, this two-step process has been proved to be
15 efficient to separately hydrolyze hemicelluloses and crystalline cellulose. Although there was
16 the hydrolysis of para-crystalline cellulose overlapping with the hemicelluloses in the first
17 stage, a relative trace amount in its production compared with that from the hemicelluloses
18 could be neglected. Therefore, these two important chemical components can be successfully
19 chased independently by this process.

Upon lignin, it was found that its monomeric products were firstly recovered before the temperature reached 230°C as shown in Fig. 3. This finding accords with the literature reporting that lignin degradation was probably achieved at the temperature below 200°C¹⁴. Among these compounds, coniferyl alcohol, coniferyl aldehyde, isoeugenol and vanillin were primarily found in the first stage as hydrolyzed monomeric lignin-derived products. It was certainly proved that all of them were derived from guaiacyl unit of softwood lignin. Apart from these monomeric lignin-derived products, the amounts of its dimeric, trimeric and higher products were estimated by subtracting the monomeric products from the whole soluble lignin, as described in our previous paper¹⁵. More detail in lignin study will be published elsewhere.

Decomposition of hydrolyzed products from hemicelluloses and cellulose

After the hydrolysis of hemicelluloses and cellulose discussed above, the hydrolyzed products are further decomposed if the treatment is prolonged by hot-compressed water. Generally, they are further decomposed by dehydration and fragmentation reactions²⁴. Since the decomposed products would lead to an inhibitory effect on later fermentation process for ethanol production²⁵, it is important to know its decomposition pathway.

Fig. 4 shows the obtained dehydrated products of 5-HMF, furfural and levoglucosan. Generally, 5-HMF is considered as a dehydrated product from hexoses such as glucose, mannose and galactose, while furfural from pentoses such as xylose and arabinose^{14,21,26}.

1 Levoglucosan, particularly, is a dehydrated product from glucose²¹. As softwood
2 hemicelluloses are composed of both hexoses and pentoses, not only furfural but also 5-HMF
3 and levoglucosan were possibly found in the first stage. Nonetheless, fewer amounts of them
4 were obtained, compared with the ones in the second stage. This is due to more severe
5 conditions in second stage that can enhance the degree of dehydration.

6 However, furfural was unexpectedly produced during the second stage. The same
7 finding has been realized in our previous work¹⁵. Therefore, this obtained result can confirm
8 that furfural is not only produced from pentoses but also from hexoses such as glucose under
9 the hot-compressed water treatment. The possible formation of furfural without pentoses is
10 possibly via 5-carbon ketoses pathway as proposed in the literature²⁷.

11 In Fig. 5, methylglyoxal and glycolaldehyde can be seen in both first and second stages,
12 while erythrose in the second stage only. Therefore, at the first stage, pentoses such as xylose
13 and arabinose from xylan hemicellulose would be decomposed to glycolaldehyde and
14 glyceraldehyde, and then glyceraldehyde was dehydrated to methylglyoxal as observed in
15 glyceraldehyde pathway of hexose fragmentation²⁸. However, one molecule of pentose can
16 give one glycolaldehyde and one methylglyoxal according to the carbon balance. Therefore,
17 the ratio among their weight percentages should be approximately 1:1. However, it is obvious
18 that much more methylglyoxal was produced. Thus, there should be another source for
19 methylglyoxal production. Undoubtedly, hexoses such as glucose, mannose and galactose,

from glucomannan hemicellulose could be fragmented into methylglyoxal produced via glycerinaldehyde/dihydroxyacetone pathway in hexose fragmentation^{21,29}, leading to the additional amount of methylglyoxal over the glycolaldehyde.

At the second stage, on the other hand, glycolaldehyde and erythrose formation via retro-aldol condensation^{21,30} occurred in glycolaldehyde/erythrose pathway, while methylglyoxal was produced via glycerinaldehyde/dihydroxyacetone pathway in hexose fragmentation as mentioned above. However, under the conditions applied, glycerinaldehyde and its isomerized dihydroxyacetone in glycerinaldehyde pathway were not detected in either stage. This would be because the dehydration reaction of glycerinaldehyde is so fast consecutively to methylglyoxal and/or organic acids²⁸.

Production of organic acids

Organic acids are considered as decomposition products of the dehydrated and fragmented compounds²¹. As in Fig. 6, the produced organic acids are glucuronic acid, acetic acid, lactic acid, glycolic acid and formic acid. However, glucuronic acid was the hydrolyzed product from the glucuronic acid residue in the minor softwood hemicellulose, arabino-4-*O*-methylglucurono-xylan, while acetic acid found in the first stage mostly came from hydrolysis of acetyl group in the major one, *O*-acetyl-galactoglucomannan. Acetic acid in the second stage, on the other hand, must be a result of the decomposition of cellulose

and/or lignin^{15,31}. In addition, the production of lactic acid, glycolic acid and formic acid were found in both stages of the treatment, thus, indicating that the decomposition of dehydrated and fragmented compounds took place. Acrylic acid and levulinic acid were not detected under the conditions applied.

Overall compounds produced from hemicelluloses, cellulose and lignin

Table 2 summarizes the yields of the produced compounds from hemicelluloses, cellulose and lignin separately as treated by two-step hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min. The results clearly shows that the water-soluble portion contained 40.14(40.21-0.07)% hydrolyzed products as in forms of various saccharides, uronic acid and acetic acid (19.49(=19.24+0.25)% from hemicelluloses and 20.65(=2.95+17.77-0.07)% from cellulose), and their decomposed compounds (dehydrated and fragmented compounds plus organic acids) to be 8.69(=4.06+3.88+0.68+0.07)%, while lignin-derived products were 20.36(=13.90+6.46)%. Apart from these, 18.57% unidentified products in the water-soluble portion were obtained and considered as unknowns. On the other hand, the water-insoluble residue was only 12.24% composed entirely of lignin.

The hydrolyzed saccharides from *O*-acetyl-galactoglucomannan hemicellulose included glucomanno-saccharides, such as mannose, glucose and oligomeric glucomannan, galactose

and acetic acid, while those from arabino-4-*O*-methylglucuronoxylan hemicellulose were xylose, xylo-oligosaccharides, arabinose and glucuronic acid. On the other hand, saccharides from cellulose included glucose and cello-oligosaccharides as well as fructose, an isomerized product from glucose.

Recently, Matsunaga et al.³² treated Japanese cedar with subcritical water by using the semi-flow-type reactor. It was reported that the highest total recovered saccharide yield of about 37.26wt% on extractive-free wood flour basis could be achieved at 310-320°C, 25MPa and 65g/min flow-rate without any pretreatment to improve wettability of the wood meal. In comparison with the total hydrolyzed compounds obtained in this study (40.14wt%), the two-step hot-compressed water treatment have been proved to give higher yields of saccharides.

Since chemical composition of Japanese cedar on an extractive-free basis is 19.10, 48.32 and 32.58%, respectively, for hemicelluloses, cellulose and lignin, the total products from hemicelluloses accounted for $112.93(=(21.32+0.25)/19.10 \times 100)\text{wt}\%$, lignin $100.06(=(13.90+6.46+12.24)/32.58 \times 100)\text{wt}\%$, while cellulose $56.42(=(2.95+24.31)/48.32 \times 100)\text{wt}\%$ on each constituent basis.

In case of hemicelluloses, if the figure is calculated except a water molecule jointly, it will be of around 100%. For lignin, there existed lignin-derived products (13.90+6.46) and water-insoluble residue (12.24) so that most of the lignin-derived products were identified. On

the other hand, the residue of Japanese cedar after the two-step treatment did not show any presence of cellulose, as inspected by the Klason lignin determination. Thus, the total products from cellulose were relatively low in comparison with hemicelluloses and lignin. However, this missing part of cellulose is not known yet at this moment.

Decomposition pathway of hemicelluloses

Cellulose hydrolysis as treated by hot-compressed water has been well and adequately studied^{33,34}. In our previous work¹⁵, moreover, its decomposition pathway has been reported. However, there is no hydrolysis pathway of softwood hemicelluloses reported in detail. In this paper, the pathways of both major and minor hemicelluloses in softwood as treated hot-compressed water were proposed independently.

Fig. 7 shows the decomposition pathway of the major hemicellulose in softwood, *O*-acetyl-galactoglucomannan. The dehydration reaction could be confirmed because dehydrated products (5-HMF and levoglucosan) from hexoses (glucose, mannose and galactose) were detected in the first stage, while the formation of the fragmentation product (methylglyoxal) could be attributed to fragmentation reaction of these hexoses as well. As the other possible fragmented product, erythrose from hexoses, on the other hand, was not detected in the first stage. It could be, therefore, inferred that the 6-carbon carbohydrates might be stable or the decomposition into erythrose might not occur at the low temperature of

the first stage. In addition, although there was a large amount of glucomanno-saccharides produced in the first stage, glucose was formed very less, as described earlier. As a result, fructose was scarcely produced at this stage, as seen in Table 2. The other reason that might explain this low fructose yield is that the isomerization reaction preferentially proceeds in a higher temperature under the studied conditions. By this reason, thus the isomerization of glucose to fructose was not included in this proposed pathway of the hemicellulose.

The decomposition pathway of the minor hemicellulose in softwood, arabino-4-*O*-methylglucuronoxylan, is shown in Fig. 8. In the same manner, its dehydration and fragmentation reactions were confirmed by the production of furfural, glycolaldehyde and methylglyoxal from the constituent pentoses in this hemicellulose such as xylose and arabinose.

Concluding remarks

The two-step semi-flow hot-compressed water treatment was applied to the hydrolysis of Japanese cedar. It was found that the two-step treatment was proved to be effective to obtain hydrolyzed products from hemicelluloses and cellulose, separately. As a result, the predominant softwood hemicellulose, *O*-acetyl-galactoglucomannan (glucomannan) was hydrolyzed to be glucomanno-saccharides, such as mannose, glucose and oligomeric glucomannan, galactose and acetic acid, while the minor component of softwood

1 hemicelluloses, arabino-4-*O*-methylglucuronoxylan (xylan), to be xylo-saccharides, such as
2 xylose and xylo-oligosaccharides, glucuronic acid and arabinose. The para-crystalline
3 cellulose was found to be readily hydrolyzed at 230°C/10MPa into cello-saccharides, such as
4 glucose and cello-oligosaccharides, in the first stage, starting from the retention time of 10
5 min. This result was evidently attributed to the same finding in our previous study on
6 Japanese beech¹⁵ that there existed also the hydrolysis of para-crystalline cellulose at the same
7 conditions in the first stage, while crystalline cellulose was hydrolyzed at 280°C/10MPa. In
8 addition, glucose, hydrolyzed product of cellulose, was found to be isomerized to fructose and
9 mannose during the second stage. Upon lignin, it was decomposed at even lower temperature
10 to produce coniferyl alcohol and its aldehyde and fragmented compounds. These lines of
11 information are very important and useful to utilize efficiently various kinds of lignocellulosic
12 materials for biochemicals and biofuels.

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4 fermentation and hydrogenolysis”, and under the partially financial support by the Global
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6 highly acknowledged.

Figure legends

- Fig. 1 Hydrolyzed products from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min. Left axis corresponds to treatment temperature (open circles); right axis corresponds to product yield.
- Fig. 2 Monosaccharides from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min. An arrow shows the presence of mannose in the first stage, while an inserted figure is the enlarged one of the fructose and mannose peaks in the second stage.
- Fig. 3 Lignin-derived monomeric products from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min
- Fig. 4 Dehydrated compounds produced from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min
- Fig. 5 Fragmented compounds produced from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min
- Fig. 6 Organic acids produced from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min
- Fig. 7 The proposed decomposition pathway of *O*-acetyl-galactoglucomannan in Japanese cedar as treated by two-step semi-flow hot-compressed water at

230°C/10MPa/15min and 280°C/10MPa/30min

Fig. 8 The proposed decomposition pathway of arabino-4-*O*-methylglucuronoxylan in Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min

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Table 1 Cell wall components of extractive-free Japanese cedar used in this study¹⁶

wt% on extractive-free wood flour basis				
Hemicelluloses		Cellulose	Lignin	
Glucomannan	Xylan		Klason lignin	Acid-soluble lignin
12.41	6.69	48.32	32.20	0.38

Table 2 Summary of various products produced from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min

Compound	Yield (wt%)						Total
	First stage [†]			Second stage [†]			
	Hemicell	Cellulose	Lignin	Hemicell	Cellulose	Lignin	
<u>Hydrolyzed products from hemicelluloses and cellulose</u>							
	19.24	2.95	-	0.25	17.77	-	40.21
- Glucomanno-saccharides	11.54	-	-	0.17 [*]	-	-	11.71
- Galactose	1.33	-	-	-	-	-	1.33
- Acetic acid	0.09	-	-	-	0.07 ^{**}	-	0.16
- Xylo-saccharides	5.37	-	-	0.08 [*]	-	-	5.45
- Arabinose	0.85	-	-	-	-	-	0.85
- Glucuronic acid	0.06	-	-	-	-	-	0.06
- Cello-saccharides	-	2.94 ⁺	-	-	17.05	-	19.99
- Fructose	-	0.01 [‡]	-	-	0.65 [‡]	-	0.66
<u>Dehydrated compounds</u>							
	0.22	-	-	-	3.84	-	4.06
- 5-HMF	0.10	-	-	-	3.12	-	3.22
- Furfural	0.03	-	-	-	0.22	-	0.25
- Levoglucosan	0.09	-	-	-	0.50	-	0.59
<u>Fragmented compounds</u>							
	1.63	-	-	-	2.25	-	3.88
- Erythrose	-	-	-	-	0.45	-	0.45
- Methylglyoxal	1.07	-	-	-	1.19	-	2.26
- Glycolaldehyde	0.56	-	-	-	0.61	-	1.17
<u>Organic acids</u>							
	0.23	-	-	-	0.45	-	0.68
- Lactic acid	0.05	-	-	-	0.14	-	0.19
- Glycolic acid	0.06	-	-	-	0.15	-	0.21
- Formic acid	0.12	-	-	-	0.16	-	0.28
<u>Lignin-derived products</u>							
	-	-	13.90	-	-	6.46	20.36
- Vanillin	-	-	0.06	-	-	0.05	0.11
- Isoeugenol	-	-	0.02	-	-	0.02	0.04
- Coniferaldehyde	-	-	0.15	-	-	0.05	0.20
- Coniferyl alcohol	-	-	0.81	-	-	0.11	0.92
- Dimeric, trimeric and higher products	-	-	12.86	-	-	6.23	19.09

Total	21.32	2.95	13.90	0.25	24.31	6.46	69.19
Unknowns	-	-	-	-	-	-	18.57
Water-insoluble residue							12.24

Remarks:

The term “Hemicell” refers to hemicelluloses.

† including the transient time

+ Cello-saccharides produced from hydrolysis of para-crystalline cellulose

‡ Fructose, as an isomerized product of glucose, is considered as the hydrolyzed product from cellulose.

* Glucomanno-saccharides and xylo-saccharides from the hemicelluloses incompletely hydrolyzed in the first stage; In case of glucomanno-saccharides, mannose derived from an isomerization of glucose is included as well.

** Acetic acid produced from decomposition of cellulose

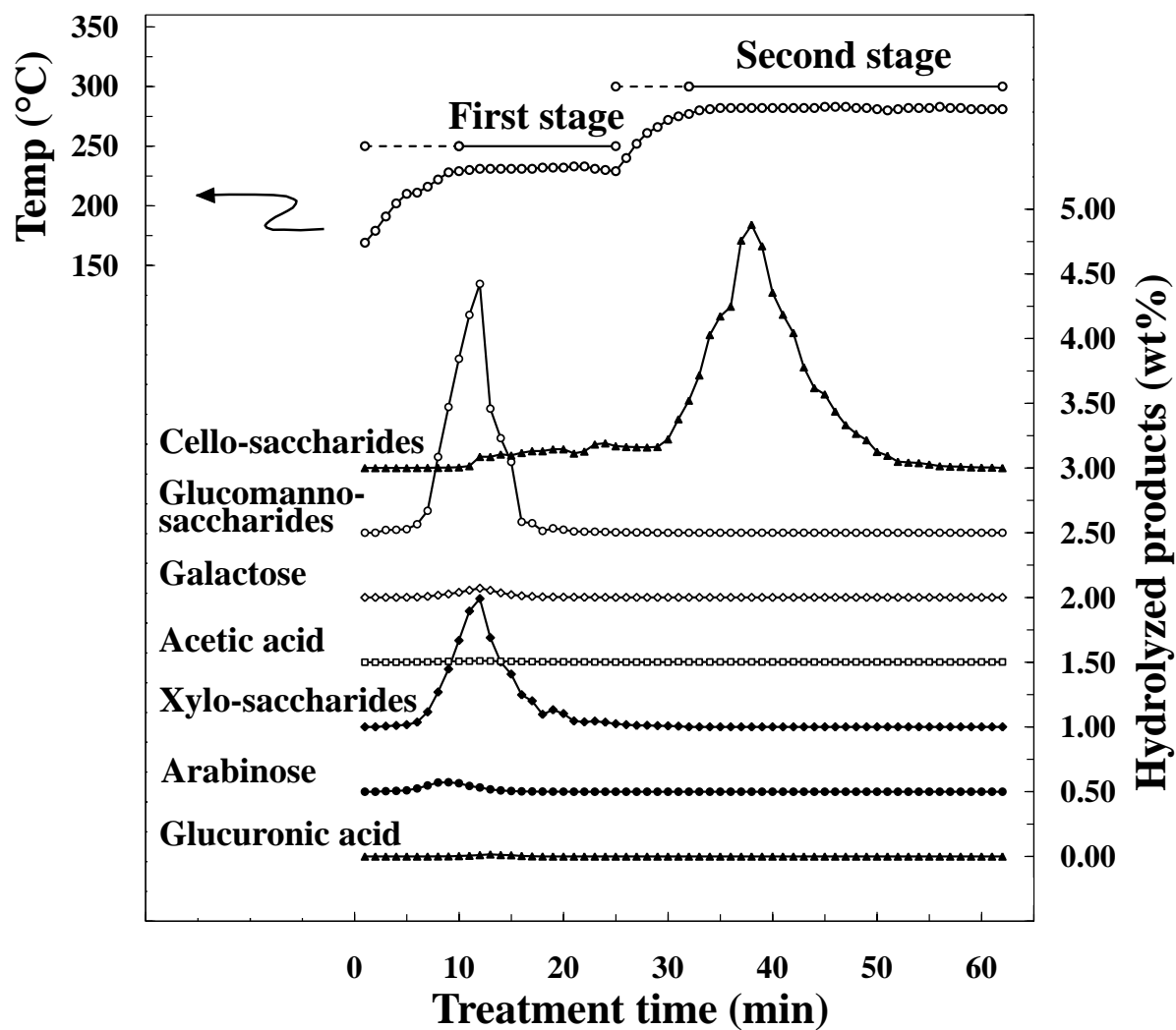


Fig. 1 Hydrolyzed products from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min. Left axis corresponds to treatment temperature (open circles); right axis corresponds to product yield.

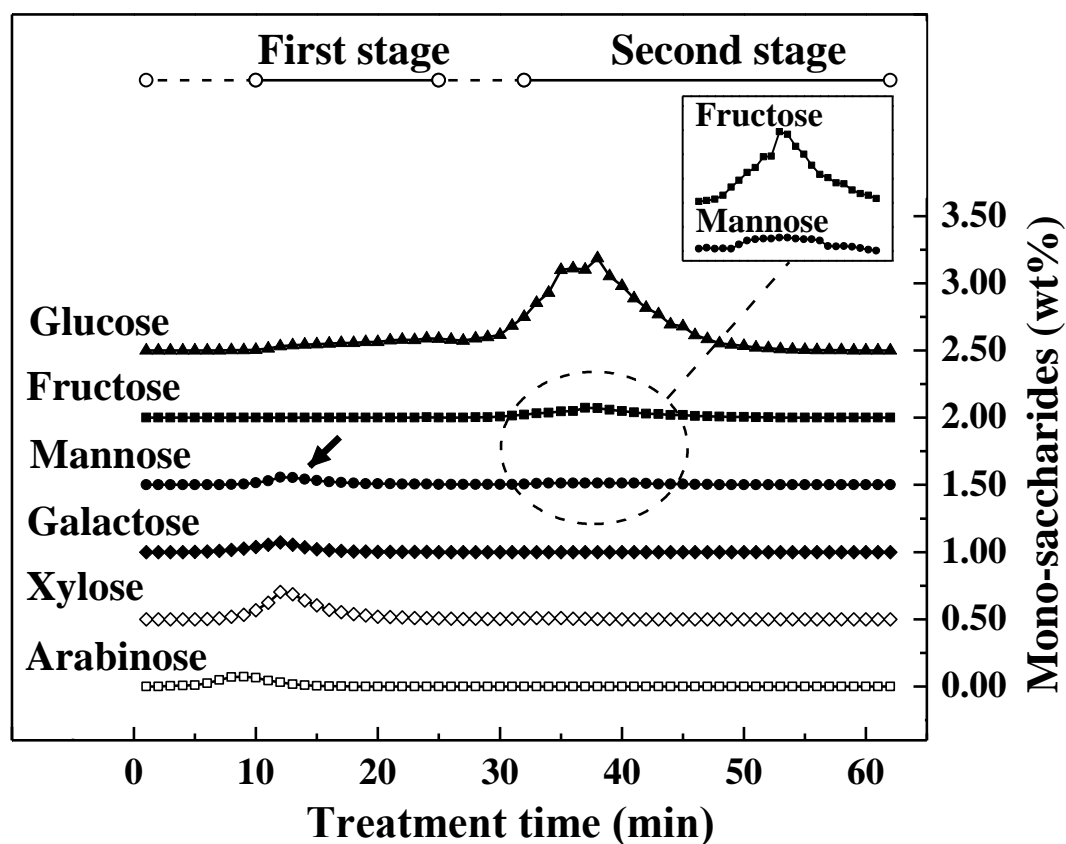


Fig. 2 Monosaccharides from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min. An arrow shows the presence of mannose in the first stage, while an inserted figure is the enlarged one of the fructose and mannose peaks in the second stage.

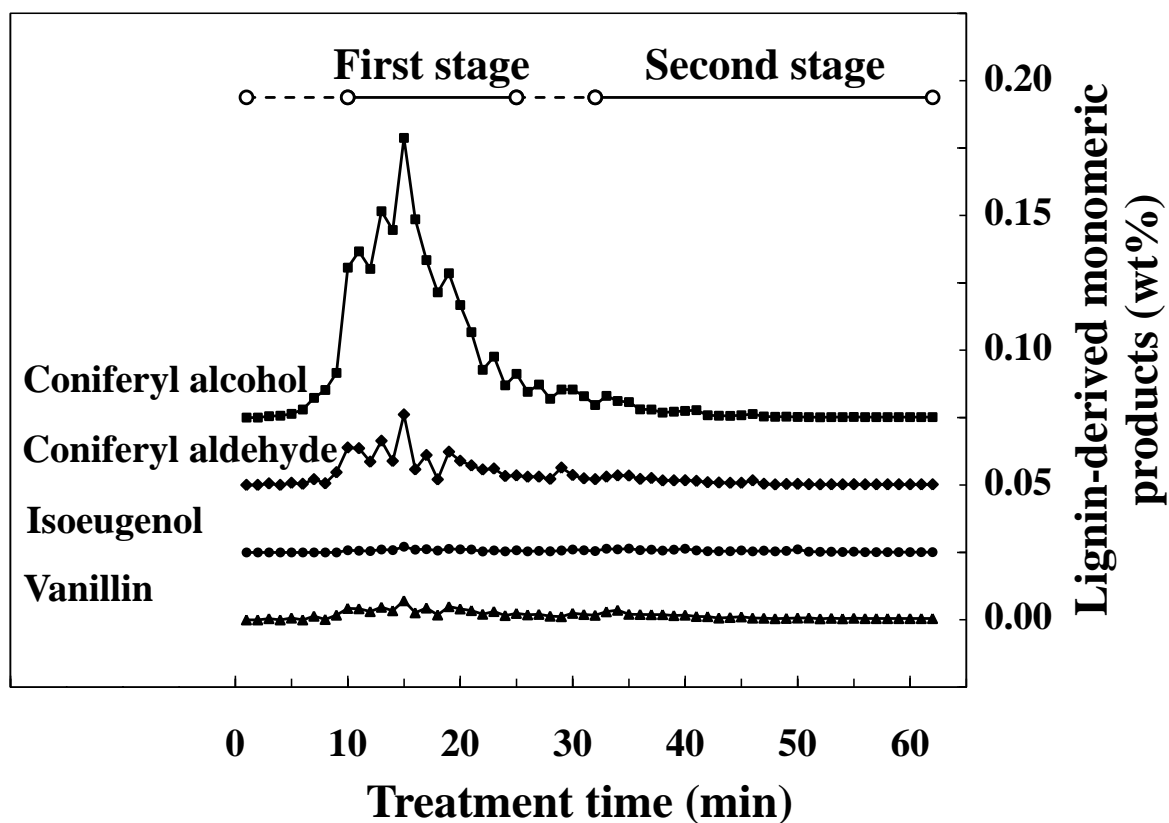


Fig. 3 Lignin-derived monomeric products from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min

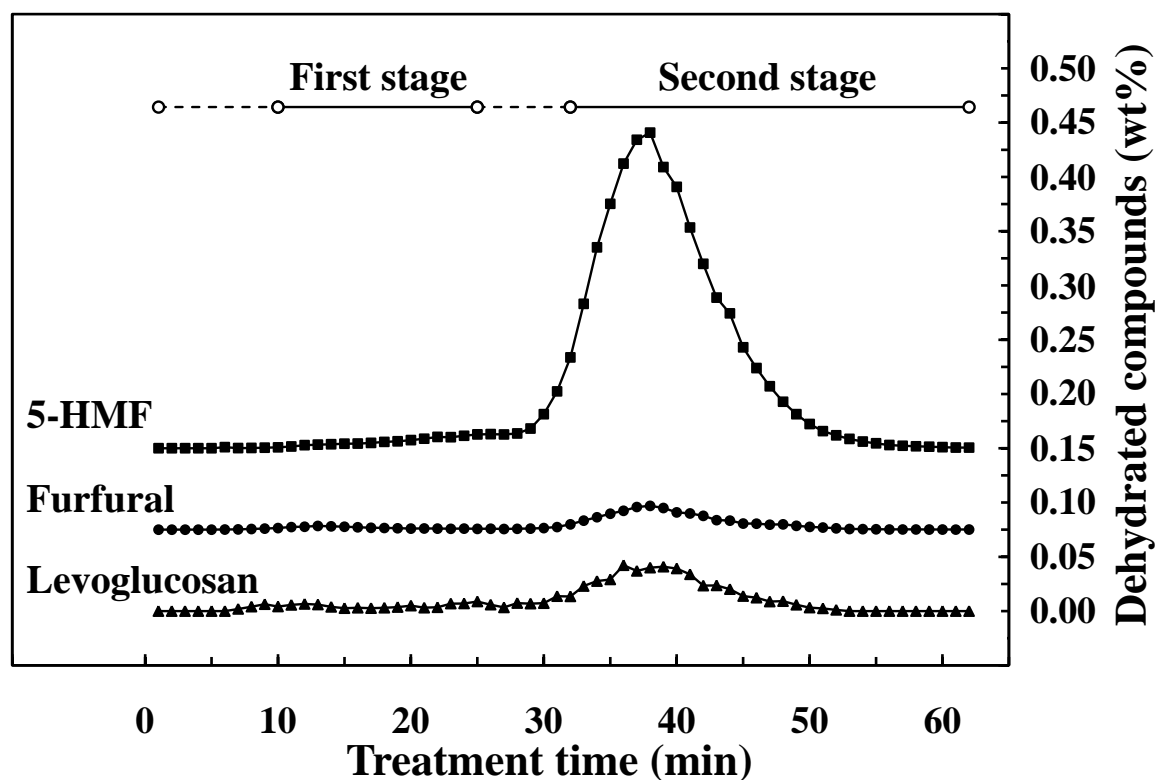


Fig. 4 Dehydrated compounds produced from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min

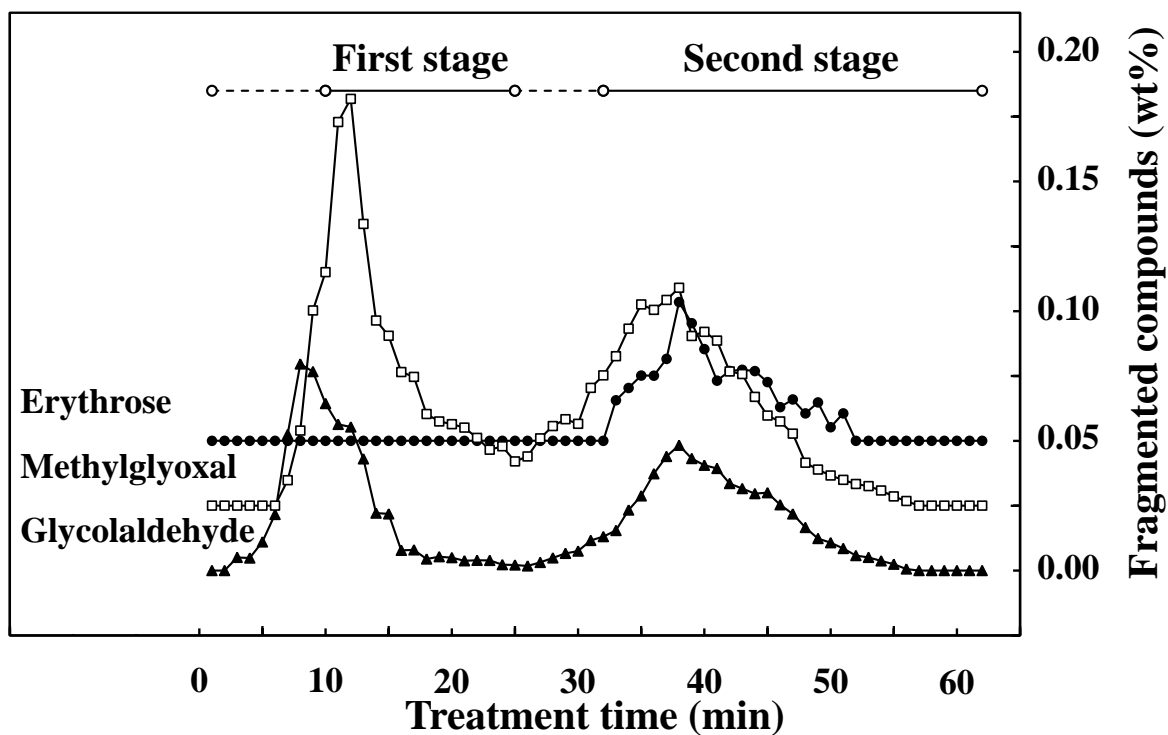


Fig. 5 Fragmented compounds produced from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min

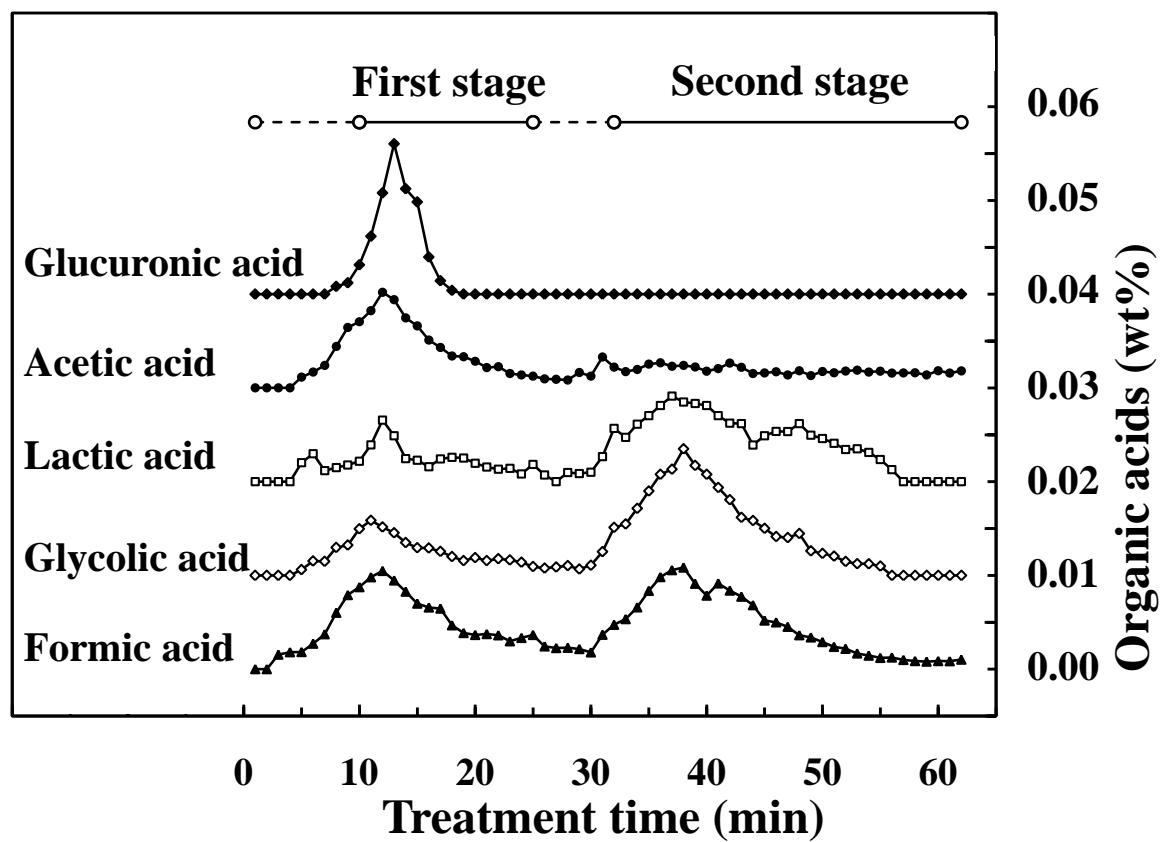


Fig. 6 Organic acids produced from Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min

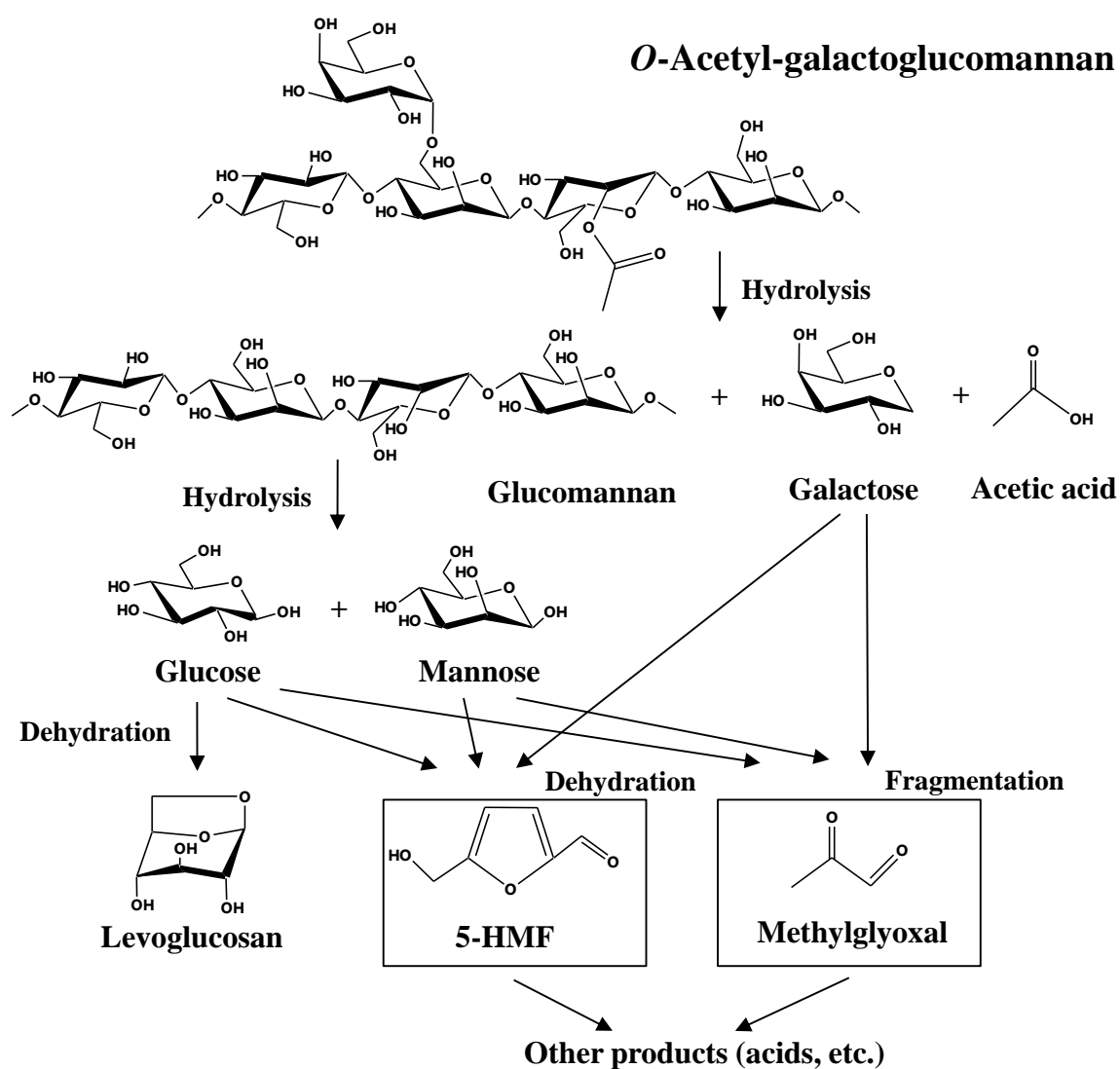


Fig. 7 The proposed decomposition pathway of *O*-acetyl-galactoglucomannan in Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min

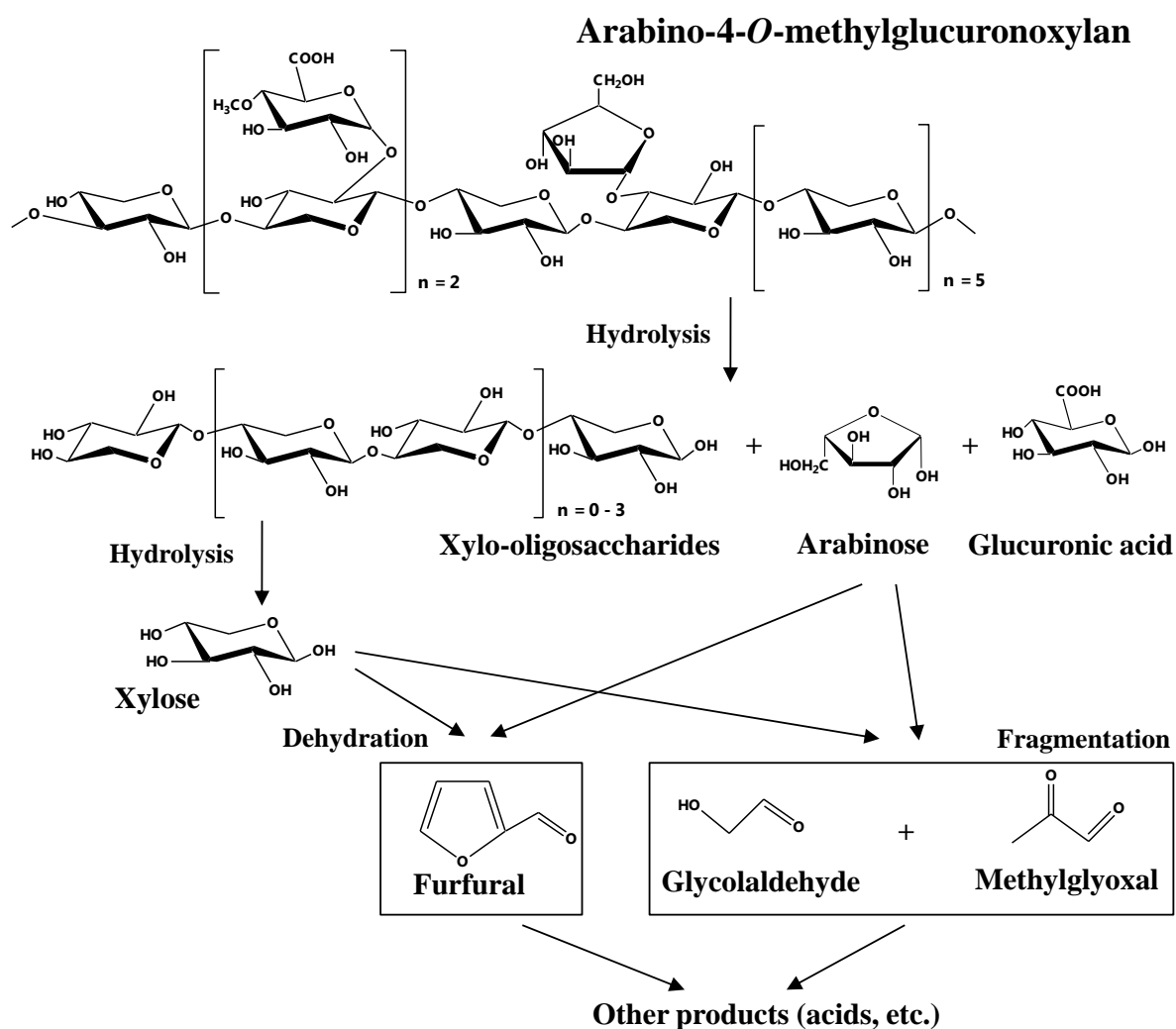


Fig. 8 The proposed decomposition pathway of arabino-4-*O*-methylglucuronoxylan in Japanese cedar as treated by two-step semi-flow hot-compressed water at 230°C/10MPa/15min and 280°C/10MPa/30min